



*Do it Right, Do it once*



# Avoiding the Epic Fail – PFAS Treatability Testing

Presented by

Michael Marley

October 17, 2018



# Outline

- ❑ Role of treatability studies in remedial design
- ❑ Treatability case studies
- ❑ Why PFAS treatability Studies?
- ❑ PFAS treatability case studies
- ❑ Collaborative treatability testing research underway
- ❑ Approximate study costs and test durations



# Why Conduct Treatability Studies?

## You Needed This:



- Certainty of success
- Select right site-specific technology
  - Determine failure mechanisms
  - Adverse reactions / byproducts
- Determine correct amount of reagents to be applied

## But What You Got Was....



# **Treatability Studies - Design**



# Example 1 – Bioremediation SOP<sup>1</sup> vs. SOA<sup>2</sup>

**Superfund Site SC: Mixed source / plume with chlorinated solvents and petroleum hydrocarbons**

## **Aerobic Biodegradation: Comparison of oxygen release products for petroleum plume**

- ❑ Evaluated oxygen release compounds on the market
- ❑ Provided vendors with site specific data and requested recommended dosing of product = SOP
- ❑ Based on responses – tested all products at MAXIMUM dosage recommended by any of the vendors\*

\* = Some vendors recommended treatability study be performed

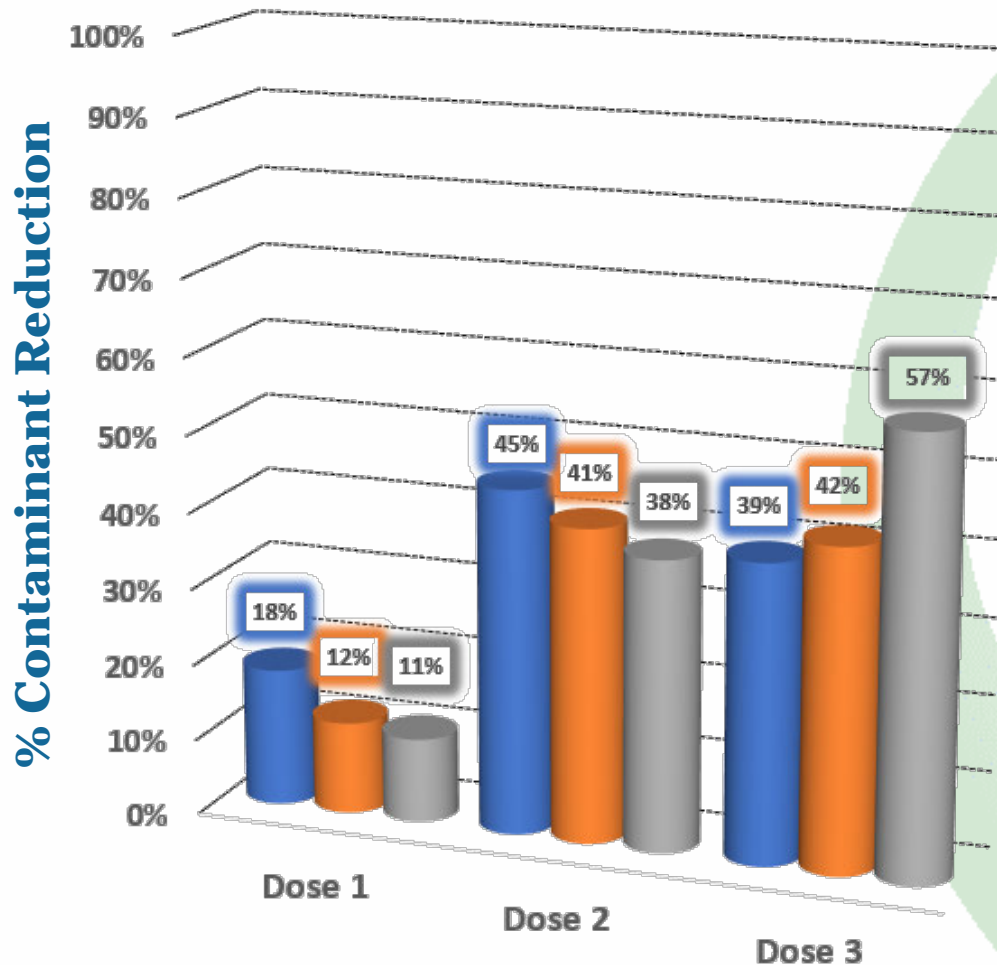
1 = State of practice

2 = State of art



# Example 1: Oxygen Release Compound Performance

Vendor Design Estimates (objective >90% Reduction with Single Dose)



All Products Failed, Even After 3 Applications at the SOP Maximum Dose Recommendation – Treatability (SOA) Identified Dose for Certainty of Success



## Example 2 –Peer Review –Diagnosis of ISCO Failure, by Others

- SOP Treatability Design using Peroxide Flawed – was “considered a success” as TCE was ND in test reactor
  - Half-life < 5hrs (from data analysis of peroxide concentration and gas generated) –not evaluated or reported
    - <5hrs half-life inadequate for oxidant distribution in the field – essentially gas generation outside well location, oxygenating the aquifer and diluting / stripping TCE
  - Loss of TCE in treatability can be accounted for by TCE vapor concentration measured in off-gas, and theoretical gas volume generated / released from mass of peroxide added
    - Gas generation was not measured / reported
  - 21 pore volumes of reagent solution used in treatability tests
    - Common lab issue
    - Not representative of field applications



# Why PFAS Treatability Studies?

- ❑ Treatability studies are perhaps even more important for PFAS than for other contaminants because:
  - Target PFAS and remedial goals are changing fast
  - Complications posed by PFAS precursors
  - Part per trillion cleanup levels
  - Potential requirement for remediation treatment trains.
  - Analytical limitations
- ❑ Each potential remedial technology requires treatability
  - Effectiveness for PFAS present
  - Byproduct formation
  - Costing for application





# Analytical Challenges

- ❑ Low detection limits required
- ❑ Cross-contamination
- ❑ Deciding which analytes to quantify of the many that exist
- ❑ Standards not available for many analytes
- ❑ Widely varying chemical/physical characteristics of PFAS
- ❑ Fast-changing regulatory requirements and analytical methods



# Common PFAS Remedial Technologies In Use



# Adsorption/Ion Exchange

(most commonplace, non-destructive, produces concentrated PFAS waste)

## ❑ Carbon-based systems

- Ex-situ activated carbon systems (GAC or PAC)
- Biochar (biomass and charcoal) – less consistent and kinetically slower?
- In-situ injectable carbon-based systems – \* gaining interest \*
- Competition with organics for sorptive sites – may require pretreatment

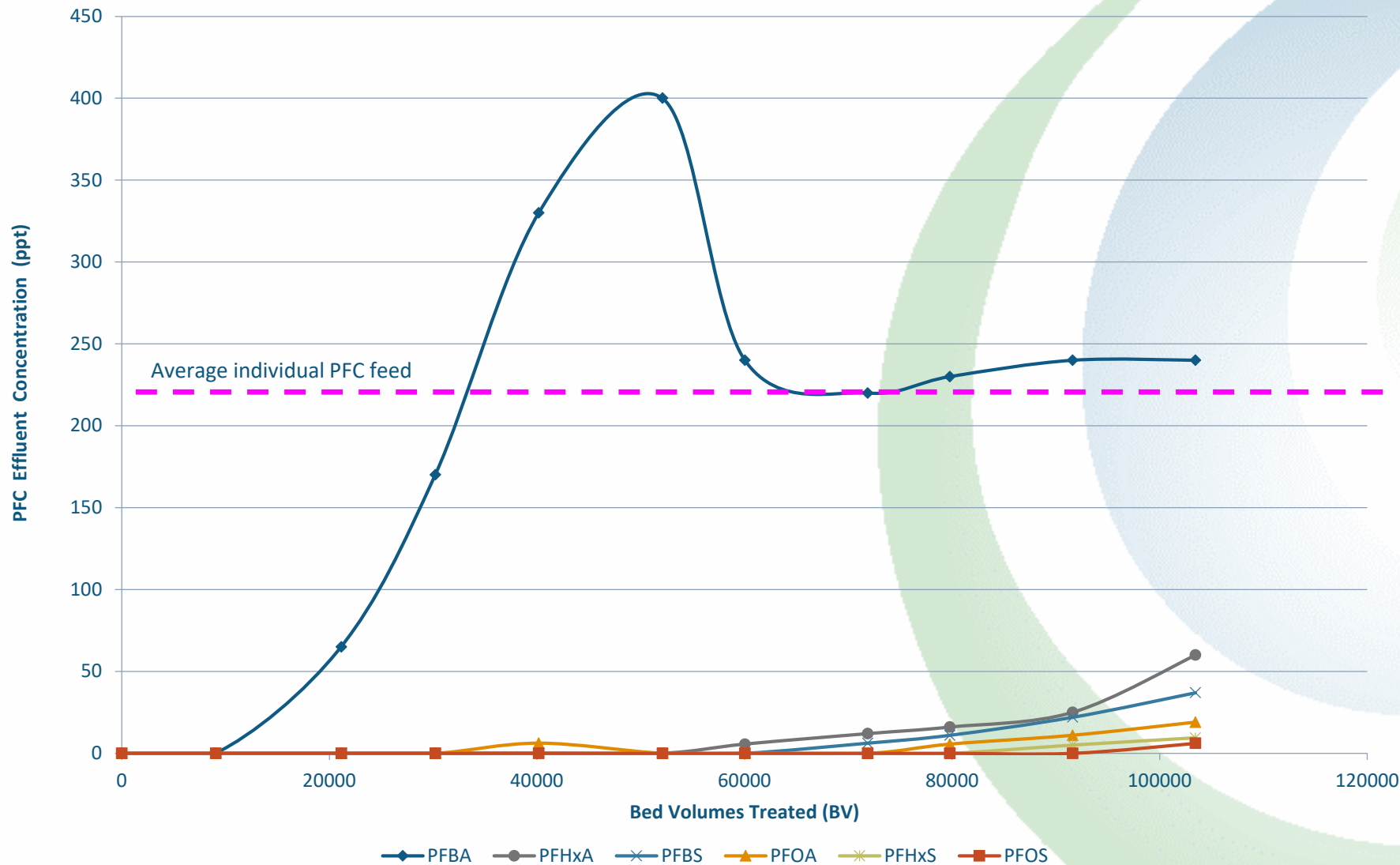
## ❑ Synthetics resins – gaining traction due to capacity/effectiveness

- Combination IX and adsorption
- Faster kinetics and higher capacities = smaller reactor size
- Higher product cost – requires site specific cost-benefit analysis
- Ongoing work on single use IX and shorter chain PFAS sorption

Treatability studies are needed



# Removal of Various PFAS using Virgin Filtrasorb



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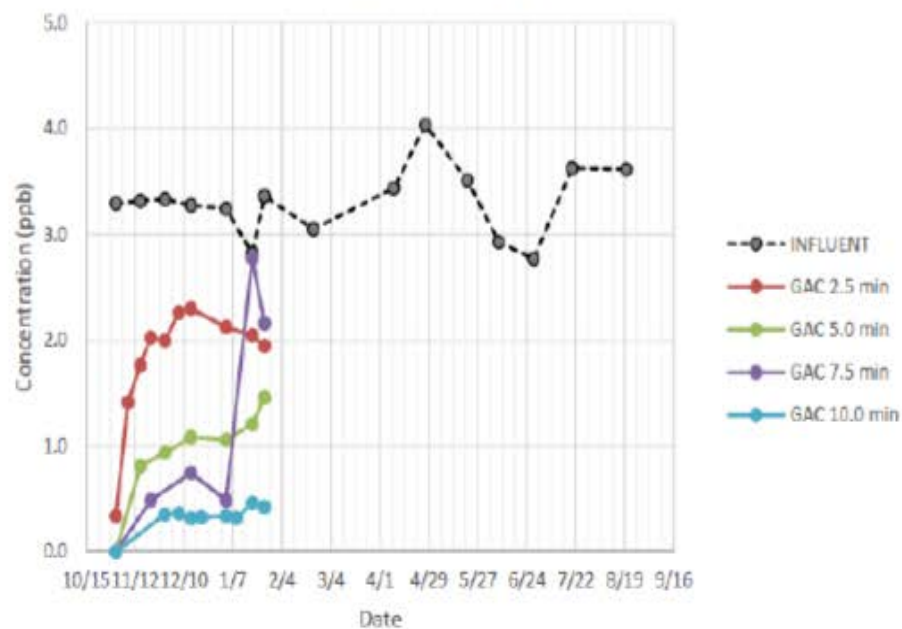


# Public water supply well (in NH) side-by-side pilot: Sorbix LC1 resin vs. Calgon F400 GAC

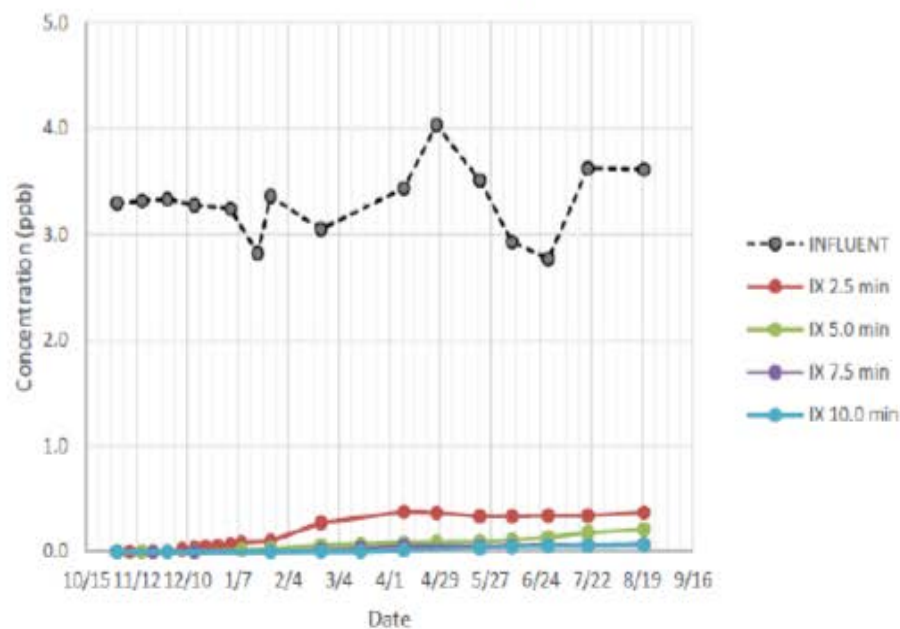


## Removal comparison – Total PFAS

### GAC - TOTAL PFAS

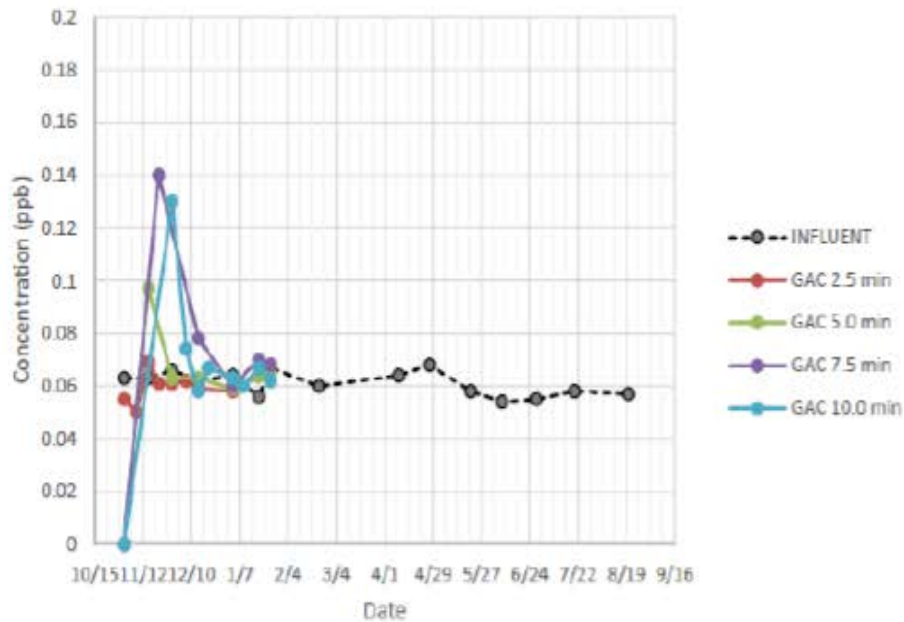


### IX - TOTAL PFAS

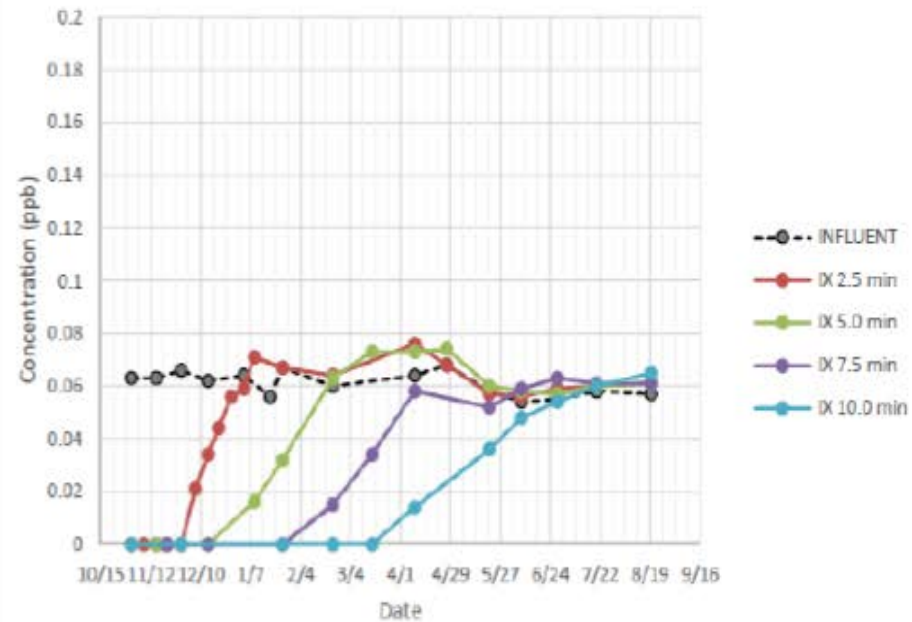


## Short chain removal comparison - PFBA

### GAC - PFBA



### IX - PFBA



# Filtration / Separation

(Also produces concentrated PFAS waste)

## ☐ Nano-Filtration (NF)

- PFAS have molecular weight cutoff (MWCO) of approximately 300 - 500 Daltons
  - >90% effective most PFAS
- Ultra and micro-filtration low effectiveness

## ☐ Reverse Osmosis

- Polymers used have spaces on the order of 100 – 200 Daltons
- >90% effective most PFAS

Treatability studies are needed





# Destructive Technologies

## □ Oxidative / reductive technologies – redox manipulation

- Can treat many of the co-contaminants
- Common theme is high energy and / or diverse reactive species needed and reaction time (e.g., electrochemical, plasma, photolysis)
- Byproducts may be a concern
  - Formation of lower C Per's with higher mobility
  - Chloride to perchlorate
  - Bromide to bromate
- PFAS range of applicability may be limited
  - Showing more promise for carboxylic' s (PFOA) than sulfonates (PFOS)
- Treatment to ppt levels may require treatment train / polishing

Treatability studies are needed



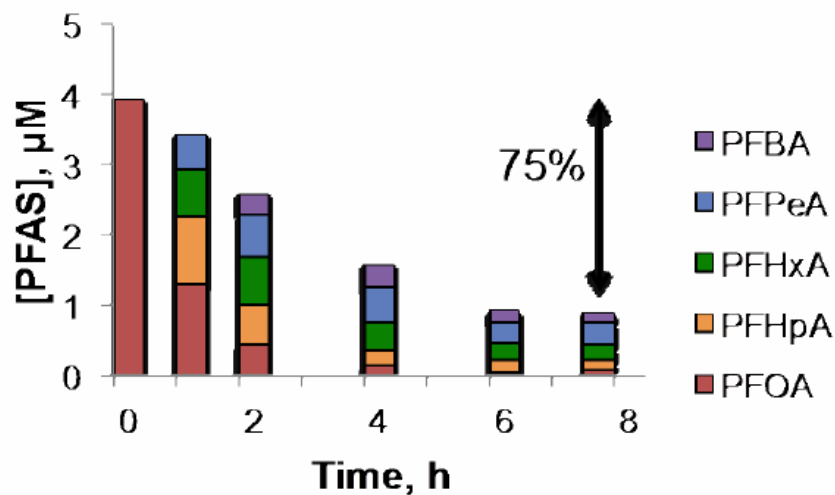
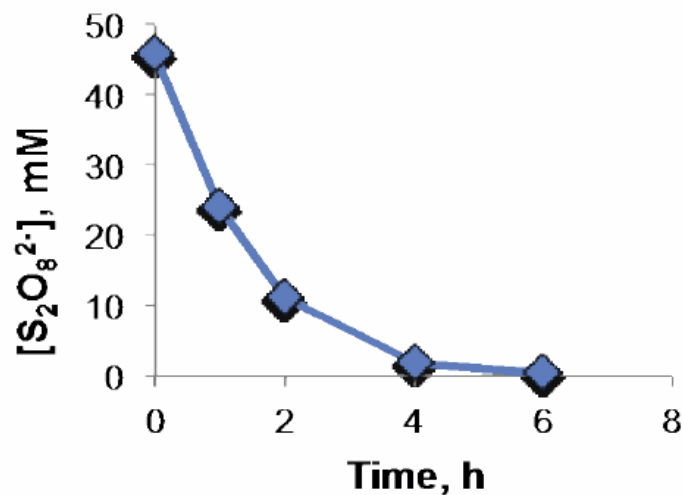
# Pretreatment of Precursors

Fighting the Unbeatable Foe: Remediation  
of Groundwater Contaminated by  
PFASs with In Situ Chemical Oxidation

Dr. David Sedlak  
University of California, Berkeley



# PFOA in Deionized Water



## Conditions:

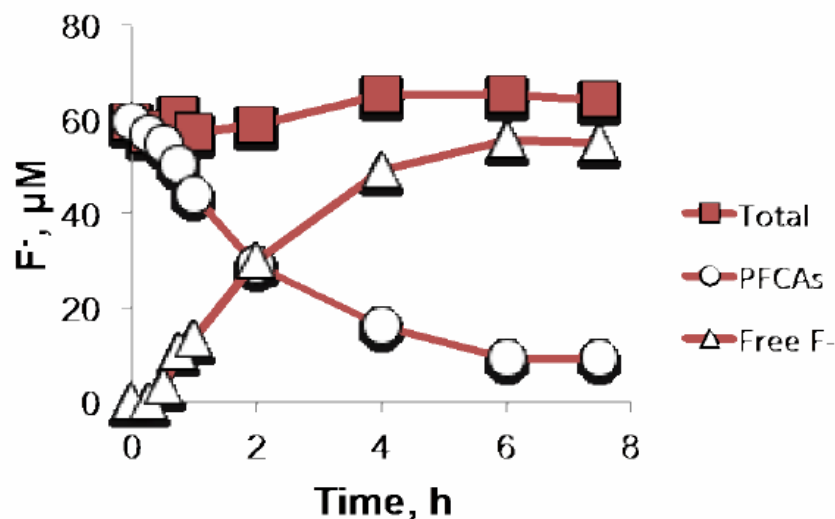
[S<sub>2</sub>O<sub>8</sub><sup>2-</sup>]<sub>0</sub> = 50 mM,

[PFOA]<sub>0</sub> = 4 μM

unbuffered (pH < 3) H<sub>2</sub>O,

T = 85° C

Bruton and Sedlak, in review



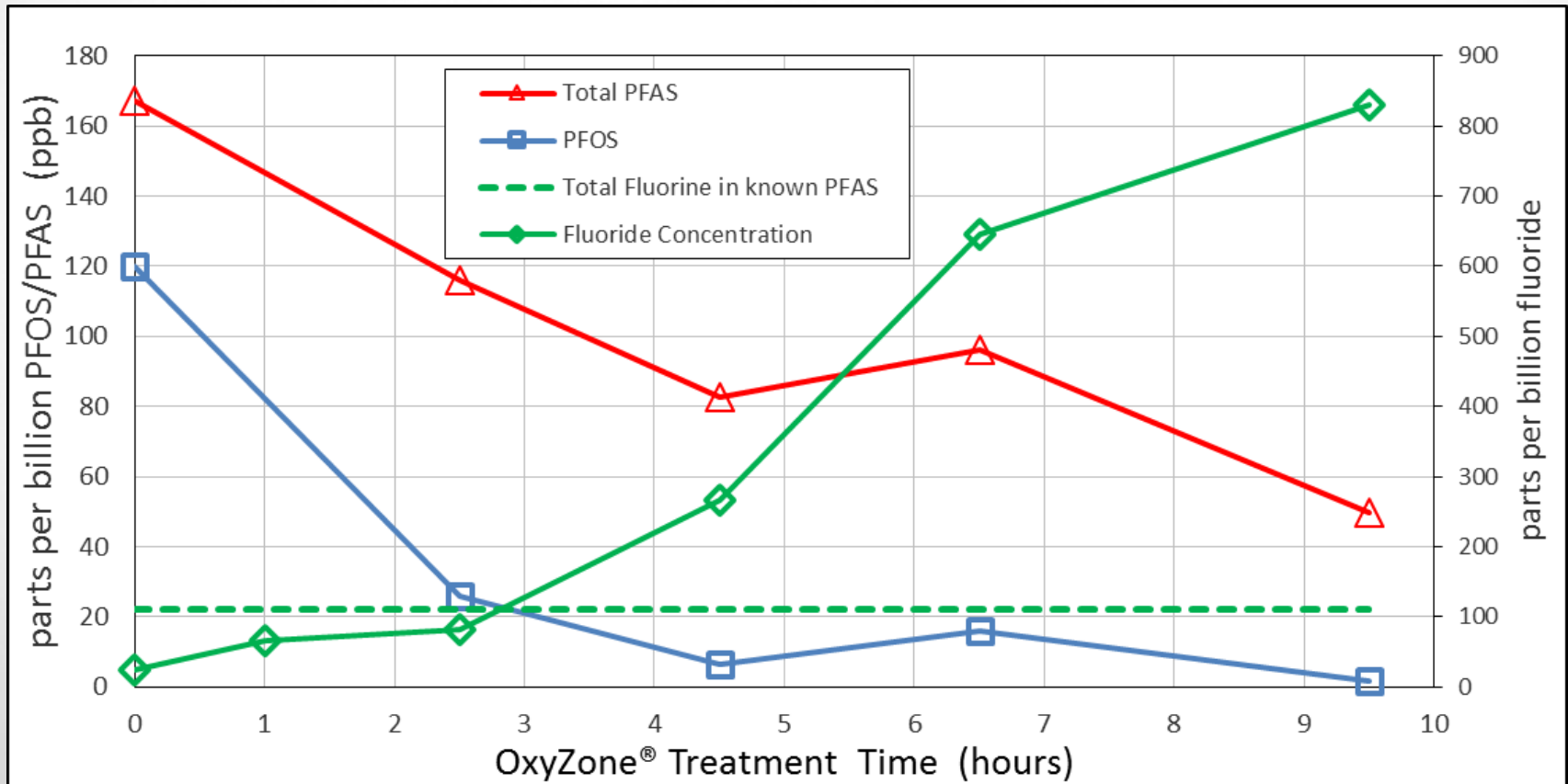
# In-House Bench Scale Treatability Testing on Groundwater (from Virginia FTA site)

## PFAS Contaminated Site GW Spiked with Additional PFOS and PFOA (6 hrs. treatment)

Specific PFAS	Initial concentration	Intermediate (3 hrs.) concentration	Final (6 hrs.) Concentration	Net Change
PFOS: (8 carbon sulfonate)	138 ppb	25 ppb	3 ppb	95% decrease
PFOA: (8 carbon acid)	33 ppb	22 ppb	6 ppb	97% decrease
PFHpS: (7 carbon sulfonate)	7 ppb	4 ppb	0.4 ppb	97% decrease
PFHpA: (7 carbon acid)	6 ppb	< 0.4 ppb	< 0.4 ppb	67% decrease
PFHxA: (6 carbon acid)	15 ppb	43 ppb	30 ppb	net increase
PFHxS: (6 carbon sulfonate)	68 ppb	99 ppb	14 ppb	79% decrease
PFPeA: (5 carbon acid)	11 ppb	< 2 ppb	< 2 ppb	91% decrease
PFBS: (4 carbon sulfonate)	9 ppb	14 ppb	10 ppb	no change
PFBA: (4 carbon acid)	3 ppb	6 ppb	5 ppb	small increase

# In-House Bench Scale Lab Results

Actual AFFF Site Contaminated Groundwater -  
 High Undetected PFAS showed 75% Fluoride Recovery



# ONGOING RESEARCH: PFAS



ALPHA  
ANALYTICAL, INC

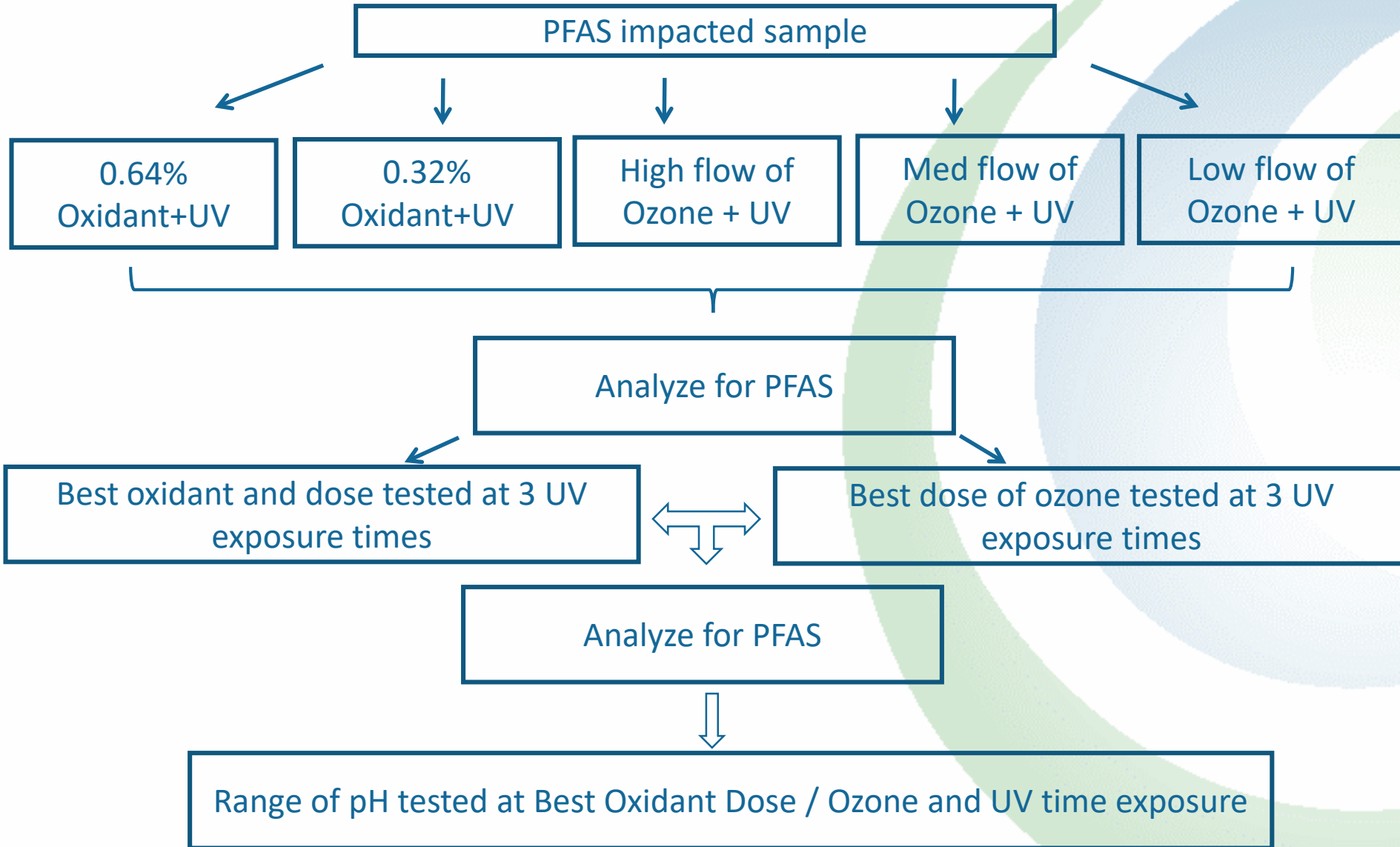


TestAmerica

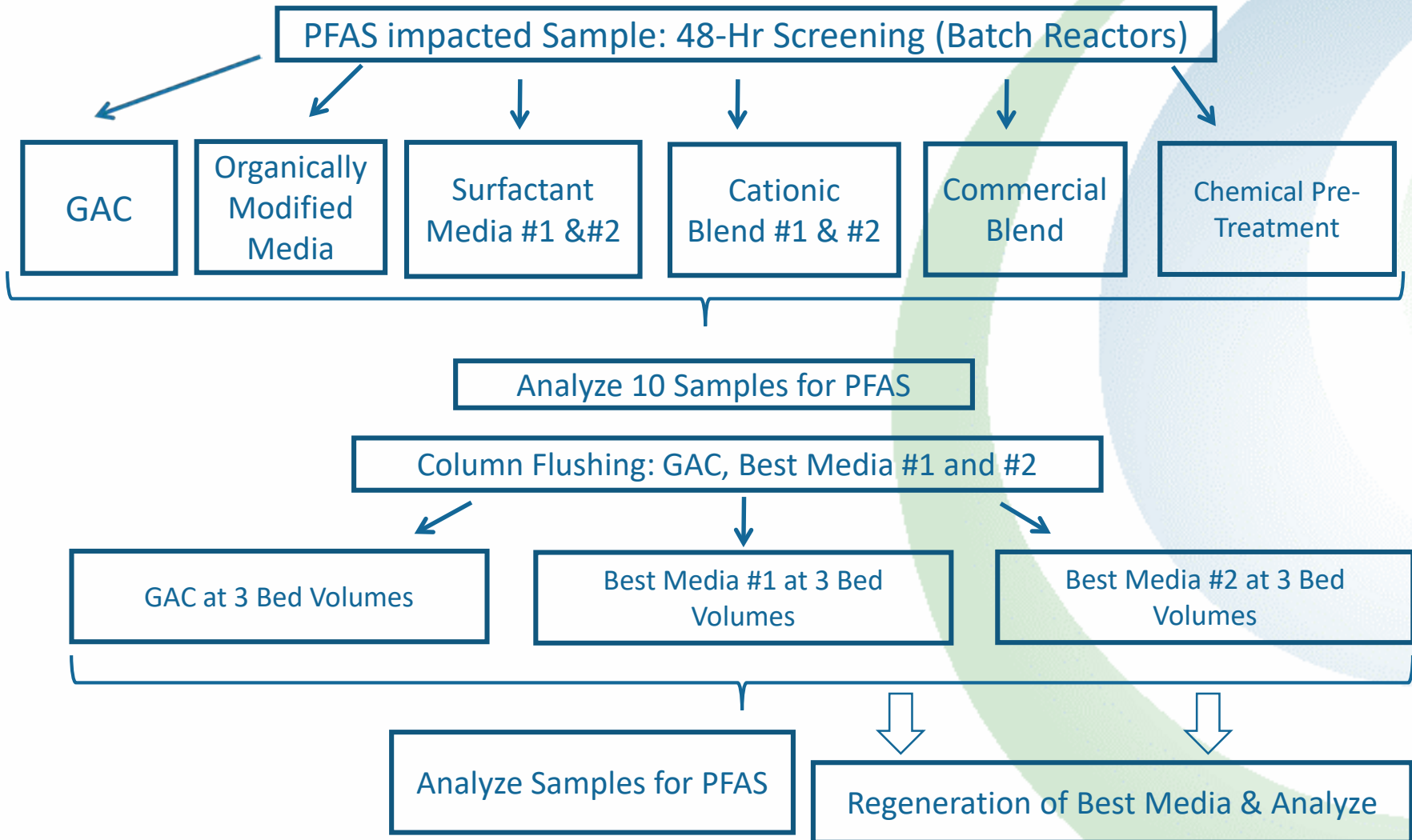
THE LEADER IN ENVIRONMENTAL TESTING



# UV/Oxidant Studies

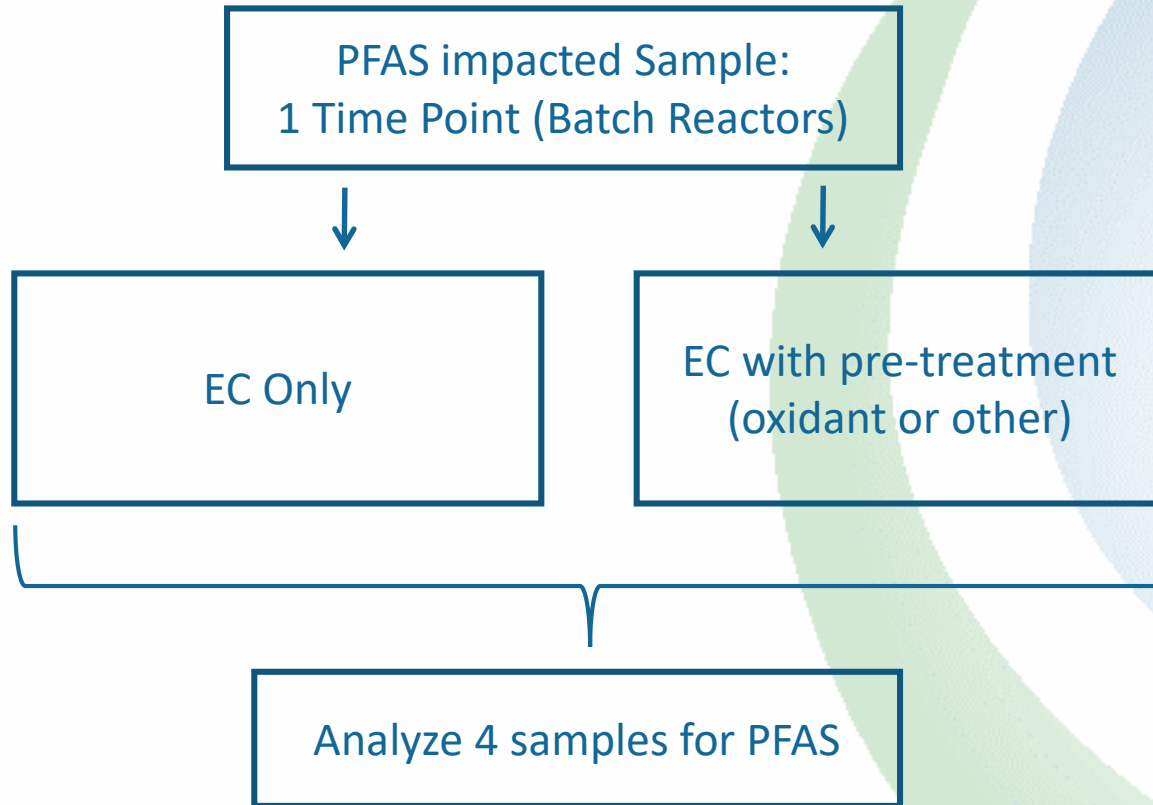


# Adsorption / Ion Exchange Studies





# Destruction (Electrochemical [EC]) Studies



# Bench Scale Testing: Duration, Media Requirements, Waste Handling, Costs

## ❑ Test Duration

- ISCO: 2 days to 8 weeks
- Bio: 2 to 6 months

## ❑ Media Requirements

- Soil: 2 to 30 pounds
- Groundwater: 1 to 20 liters

## ❑ Costs

- \$2,000 to \$50,000 or greater (function of scope and sample numbers)



# Questions?

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